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A molecular compound is a group of atoms held together by a covalent bond. Compounds made entirely of non-metals are generally molecular compounds. Carbon tetrachloride, CCl₄, is an example of a molecular compound. When naming these compounds prefixes are used to denote how many of each atom is bonded in the compound.

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In this activity you will learn the naming system for molecular compounds. Model 1 Molecular Compounds Molecular Number of Name of Compound Formula Atoms of First Atoms of Second Element Element ClF 1 1 Chlorine monofluoride ClF5 1 5 Chlorine pentafluoride CO 1 1 Carbon monoxide CO2 1 5 Carbon dioxide Cl2O 2 1

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